

SEKALOV, F.G.

Experience in the operation of the automatic communication equipment in the Vilnius long-distance telephone exchange.
Vest. sviazi 24 no.12&16-18 D '64 (MIRA 18:2)

1. Nachal'nik Vil'nyusskoy mezhdugorodnoy telefonnoy stantsii.

43475

S/205/62/002/006/001/021
E027/E410

AUTHORS: Polivoda, A.I., Sekamova, Ye.N.

TITLE: Ultraweak radiation (400 - 700 m μ) of surviving and homogenized liver tissue under normal conditions and under the influence of ionizing radiation

PERIODICAL: Radiobiologiya, v.2, no.6, 1962, 801-810

TEXT: The authors have previously observed that living mouse liver emits ultraweak radiation in the visible region (450 - 700 m μ) which can be detected with a photomultiplier, and in the present work they have used this radiation as an index of changes taking place in the tissues under the influence of ionizing radiation. The apparatus used is shown in Fig.1 and consists of a specially designed vacuum flask (2) immersed in liquid nitrogen in a specially designed vacuum flask (2) with a hole in the side through which the end of the photomultiplier (1) projects, thus avoiding loss of light due to passage through the walls of the flask. The photomultiplier has a small nontransparent photocathode, giving low noise and a high quantum yield. The condensing system consists of a quartz lens (3) and a diaphragm (4). The tissue to be examined is

Card 1/3

S/205/62/002/006/001/021
E027/E410

Ultraweak radiation ...

placed in a cuvette (6). The electronic arrangements are of standard type. The normal liver of an intact mouse exposed at operation emitted about 1000 photons/sec.cm² (average value for 135 mice), representing an energy of 2.5×10^{-6} erg/sec-1/cm⁻³ or 2.5×10^{-10} of the total energy output of the tissue. The radiation was greatly reduced 5 hours after the death of the animal or when the tissue was removed and homogenized. After an initial rise it declined during the first 4 days after exposure of the animal to a lethal dose of ionizing radiation. Liver homogenates began to radiate when the temperature was raised, attaining maximum intensity at 60°C and falling to zero at 70°C. This was attributed to thermal decomposition of peroxides. During radiation sickness the radiation of liver homogenates at 60°C rose to a maximum after 2 to 3 days and declined thereafter. There are 5 figures and 4 tables.

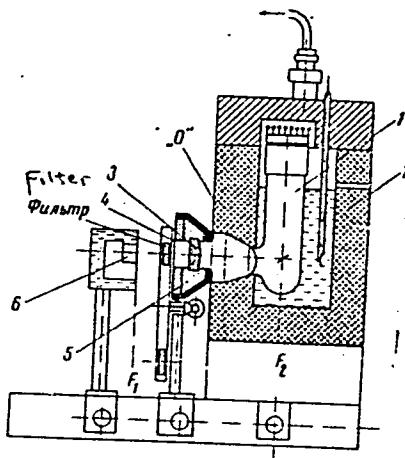
SUBMITTED: December 26, 1961

Card 2/3

Ultraweak radiation ...

S/205/62/002/006/001/021
E027/E410

Fig.1.



Card 3/3

MIKHAYLOVA, A.A.; MIKHAYLOV, L.M., POPOV, A.S.; SEKAMOVA, Ye.N.

γ irradiation of cell cultures of mammals in vitro. Radio-
biologia 5 no.4;627-628 '65. (MIRA 18:9)

VESNIKOVSKIY, V.A.; SVERDLOV, Ye.M.; TIKHONOV, V.I.

Mechanism of superweak spontaneous fluorescence of organisms.
Biophysika 8 no.1:125-127 '63. (MIR 1718)

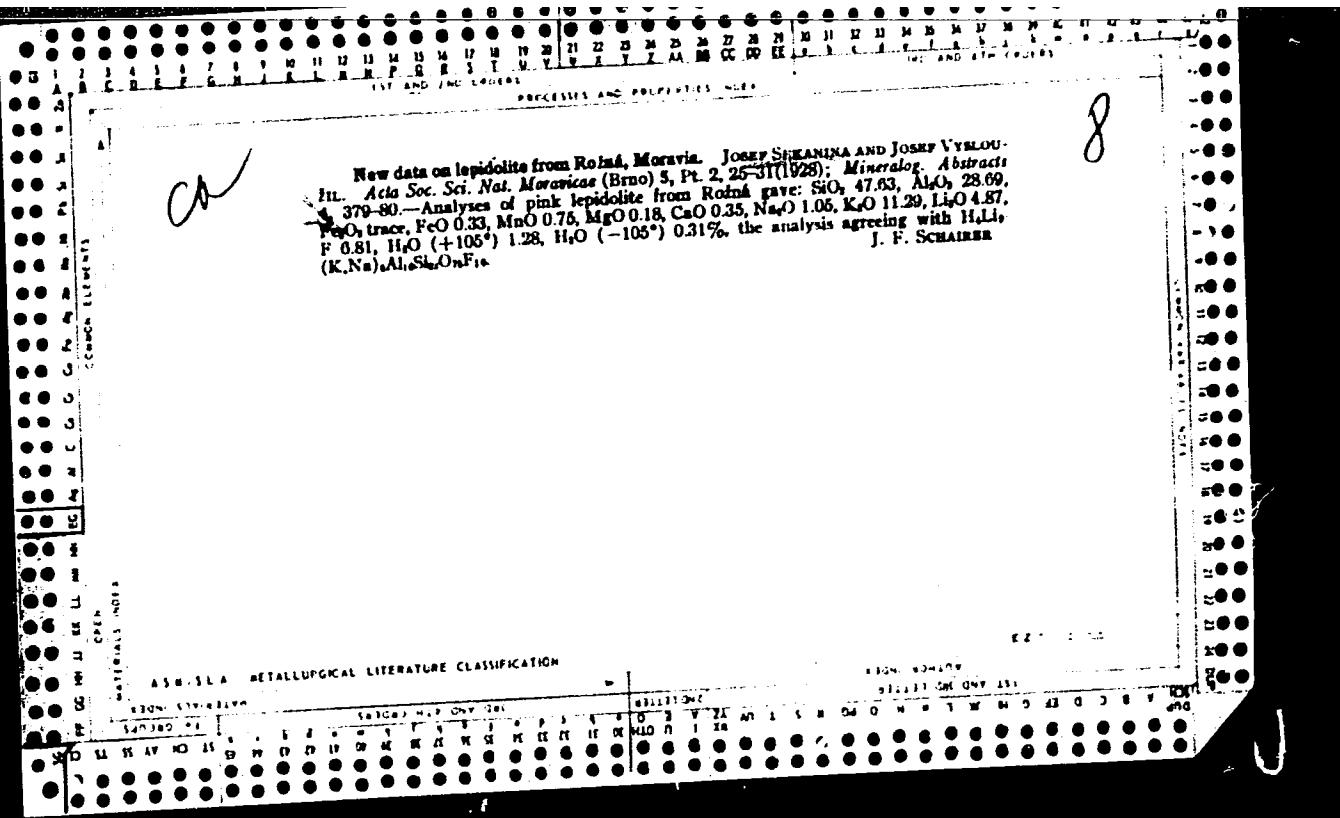
TARUSOV, B.N.; POLIVODA, A.I.; ZHURAVLEV, A.I.; SEKAMOVA, Ye.N.

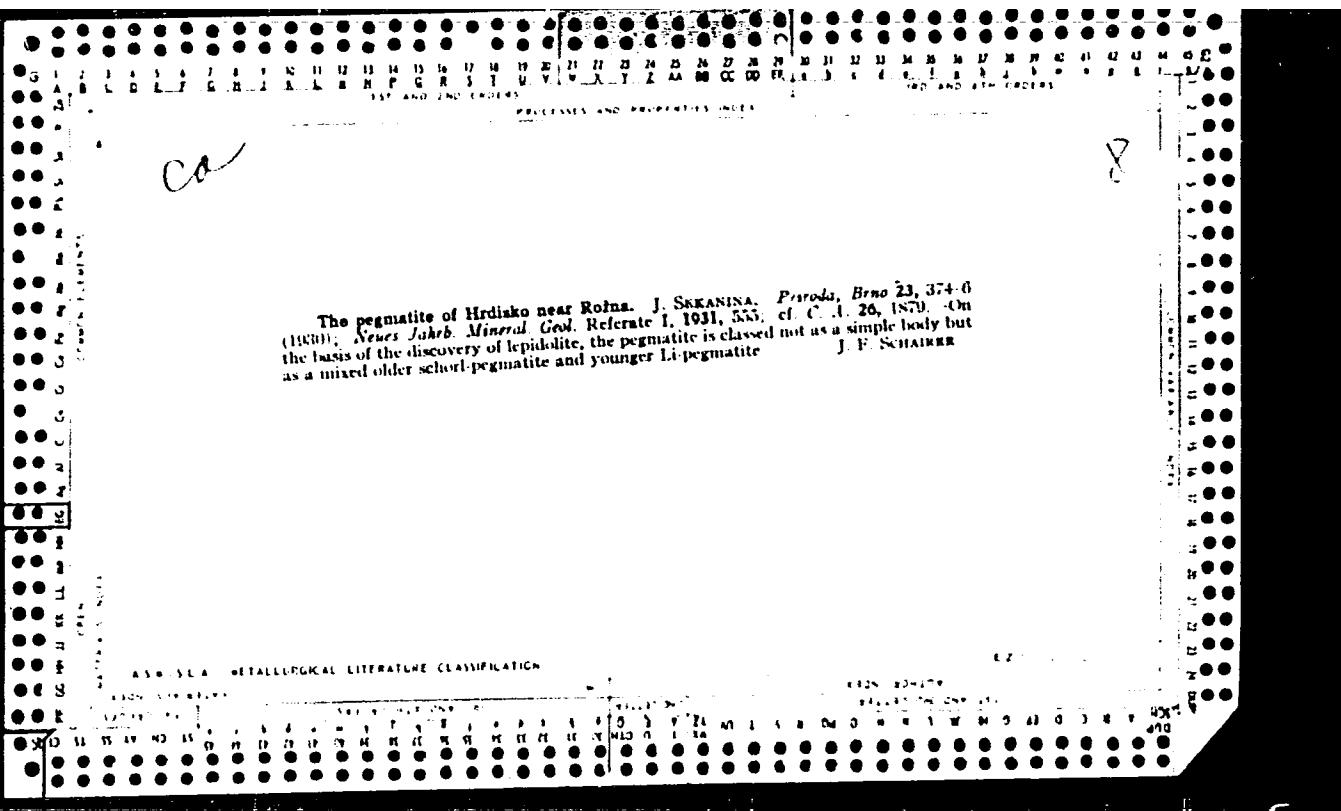
Ultraweak spontaneous luminescence in animal tissues. Teito-
logiia 4 no.6:696-699 N-D'62 (MIRA 17:3)

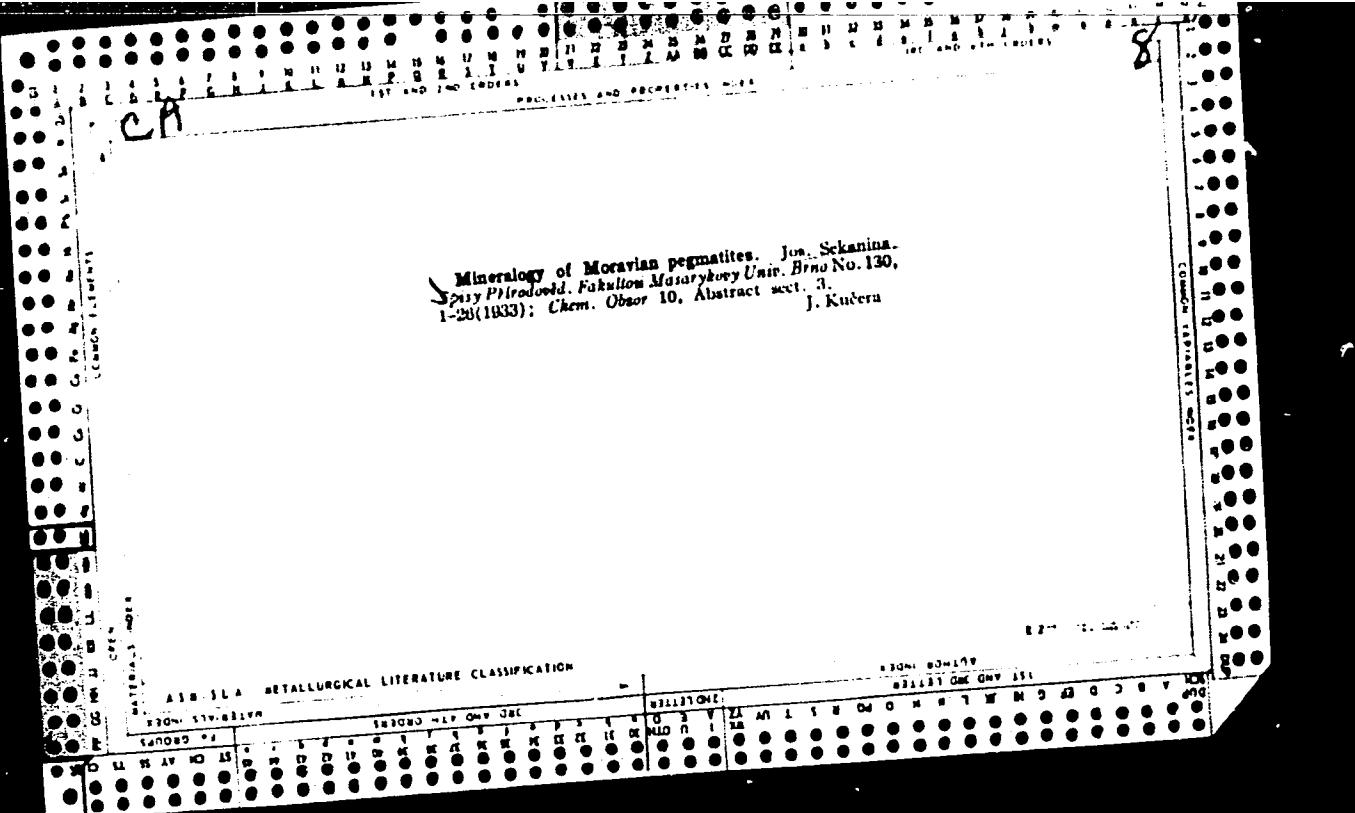
l. Akademiya meditsinskikh nauk SSSR, Moskva.

POLIVODA, A.I.; SEKAMOVA, Ye.N.

Localization of primary reactions of radiation injury in sub-microscopic tissue structures. Trudy MOIP. Otd. biol. 7:122-126 '63.
(MIRA 16:11)







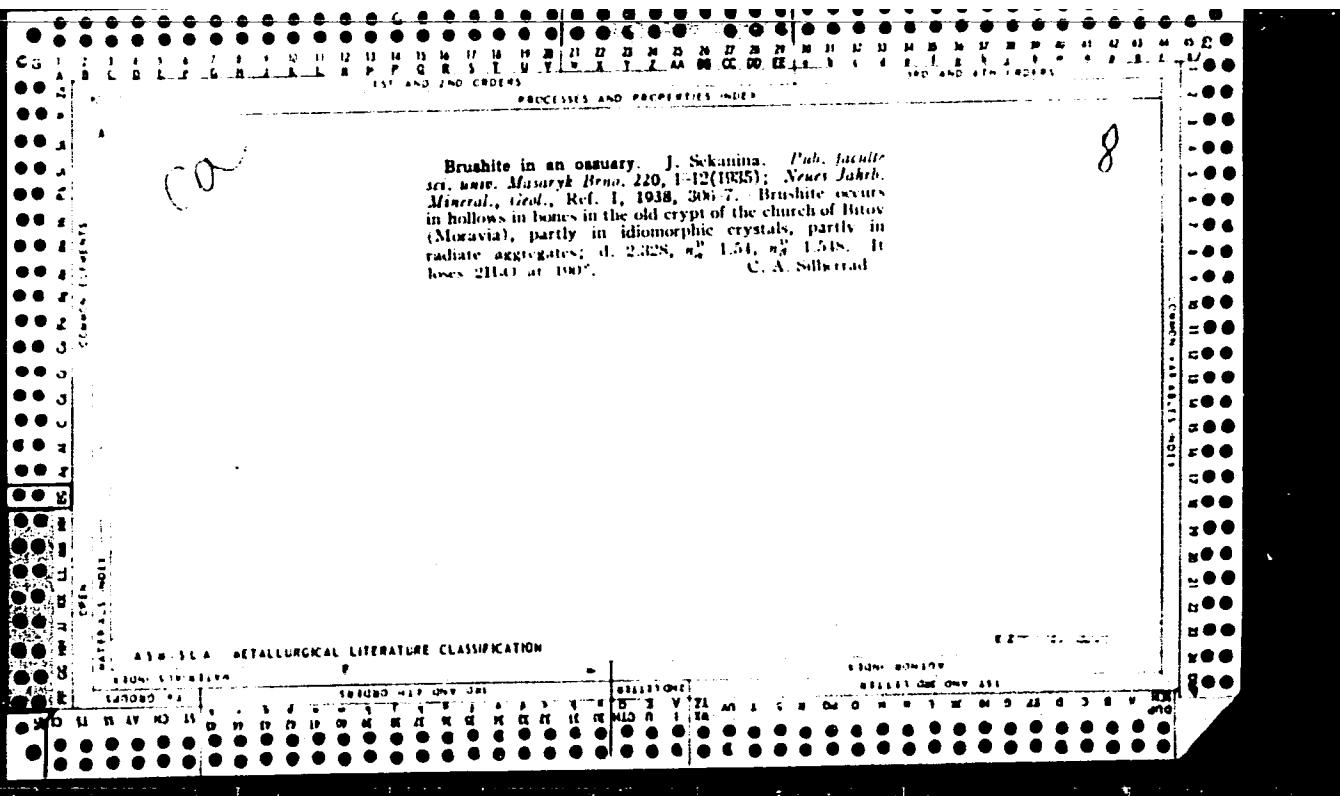
The sulfur localities of Mähren and their paragenesis
by Skámani. *Zprávy komise na přírodní výzkum Moravy a
Slezska Mineral.*, No. 6, 20 pp. (1935) (German Summary);
in *Jahrb. Mineral. Geol.*, Referate I, 1936, 408.
 α -, β - and γ -Sulfur crystals are described. Near Zarowice
S is formed by the decompos. of gypsum. J. F. S.

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ASB-SEA METALLURGICAL LITERATURE CLASSIFICATION

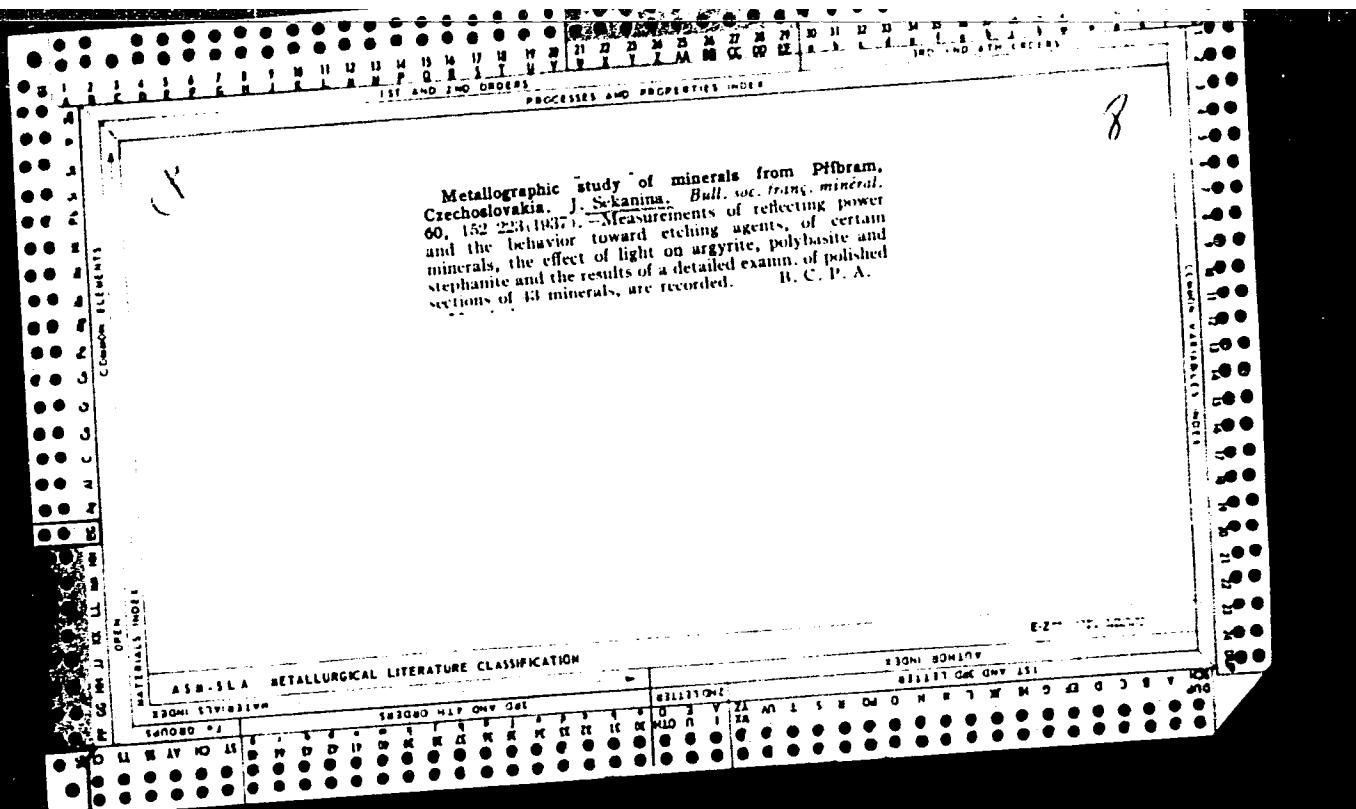
APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001547710011-4"



Origin of crystal impressions in quartzes of Hostakov
Island, Sakhalin. Shorokh Piroshnikov. Kishinev, Trans-
Siberian, 1937, no. 71, Mineralog. Zhurnat., 7, 500
(1940). Masses of zonally colored amethyst, milky and
smoky quartz occur with large cavities, whence tabular
crystals of cabelite had obviously been removed.

C. A. Silberrad

ASR-SLA METALLURGICAL LITERATURE CLASSIFICATION



Occurrence of lithium and phosphate minerals at Dobře Voda near Velké Meziříčí. Josef Šekanina (Masaryk University, Brno). *Vestn. Stat. Geol. České Rep. Českého svazu pro geologii*, 21, 209-312 [in French], 311-42 (1946). A pegmatite contained quartz, albite, perthite, muscovite, lepidolite, apatite, tourmaline, cassiterite, and anhydrite. Optical data are given for the last 3 of the minerals. Muchard Hirschler.

6

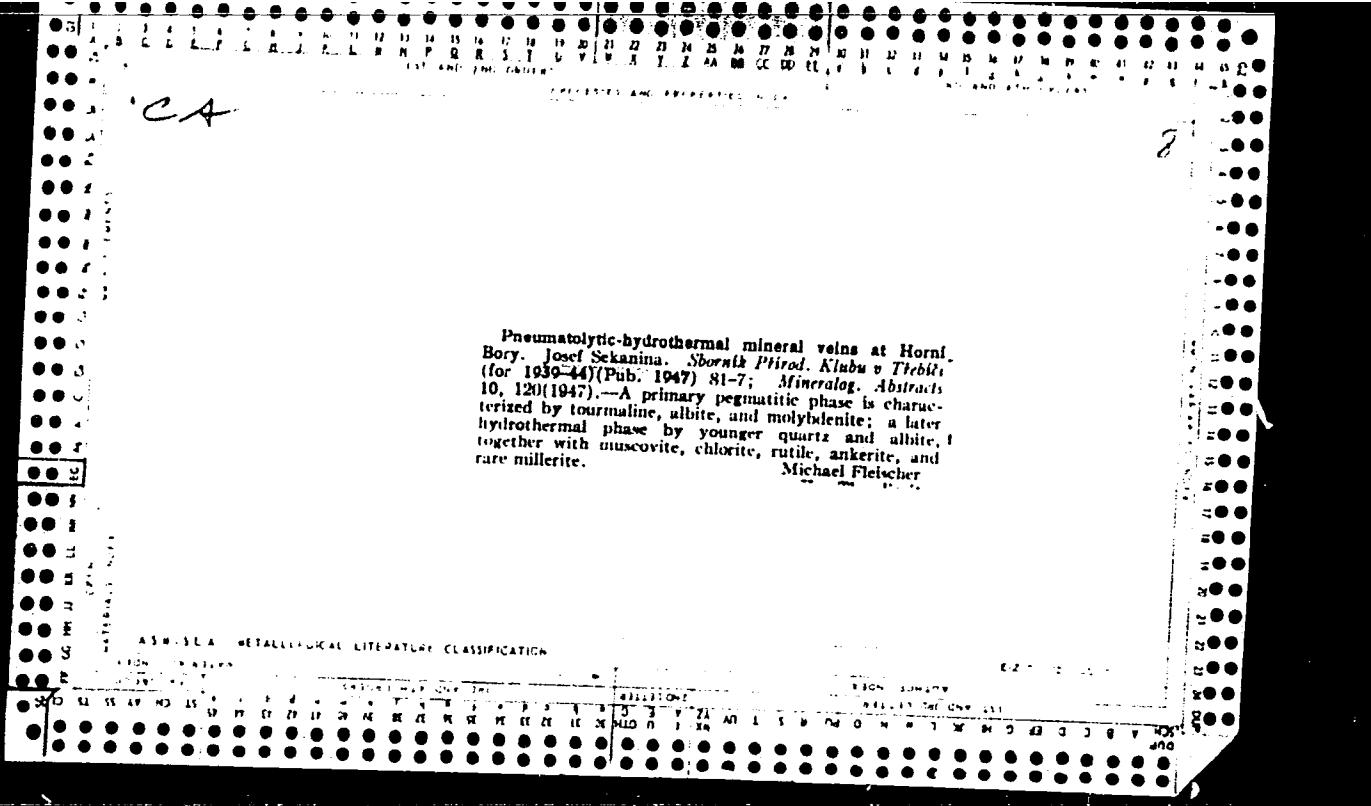
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ASB-32A METALLURGICAL LITERATURE CLASSIFICATION

APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001547710011-4"

CF
Minerals and rocks of the region between Nedvědice
and Rozna. Josef Sekanina. *Sborník Klubu Přírodovědců*
Brno 26, 99-113 (1940); *Minerolog. Abstracts* 10, 120
(1947). Axinite from metamorphic rocks, and lepidolite
and tourmaline from pegmatite are described. The Li
minerals occur in a pegmatite that cuts an older vein of
pegmatite and aplite. Partial mixing of both pegmatites
appears to have taken place at the borders. M. F.

AS-15A METALLURGICAL LITERATURE CLASSIFICATION



8

Amblygonite and neardrite at Rozná, Czechoslovakia
Josef Šekanina (Masaryk Univ., Brno, Czech). *Prace
Mářické Akad. Věd. Přírod.* 22, 211-18(1950) (French
summary). Optical and x-ray powder data are given for
these minerals. Michael Flewther

Mineralogical Summary - 8.

Minerals of the iron gossan near Nová Ves, northern Moravia. Josef Šekanina. (Masaryk Univ., Brno, Czech.). *Prace Moravské Akad. Přírodnick 23*, 107-45(1951) (French summary).--Crystallographic and optical data are given for goethite, lepidocrocite, anglesite, cerussite, pyromorphite, malachite, luarite, dundasite, smithsonite, hemimorphite, hydrozincite, chalcanthite, goislite, and greenockite. Michael Fleischer

SEKANINA, JOSEF

Mineralogie vseobecna. [Vyd. 1.] Praha, Statni pedagogicke nakl., 1953. (Ucebni texty
vysokych skol) [General mineralogy. I. Morphological crystallography. Illus.]

SO: Monthly List of East European Accessions, Vol.3, No.2, Library of Cong., Feb. 1954, Uncl.

Sekanina
S E K A N I N A ,
CZECHOSLOVAKIA / Cosmochimistry. Geochemistry. Hydrochemistry. D

Abs Jour : Ref Zhur - Khimiya, No 3, 1957, No 7857

Author : Sokanina, J.

Inst : Not given

Title : Brushite and Probable Ardoalito on Bone Remnants in Olomouts
[sic]

Orig Pub : Casop. Mineral. a Geol., 1956, Vol 1, No 1, 50-53

Abstract : Two varé phosphates, brushite ($\text{CaHPO}_4 \cdot 2\text{H}_2\text{O}$) and ardoalito ($\text{CaHPO}_4 \cdot \text{CaSO}_4 \cdot 4\text{H}_2\text{O}$), found on decomposing hyman bones in the Santa Clara monastery, are described. Brushite occurs in the form of incrustations and aggregates of fine (0.5 mm) crystals of elongated loaf-like form. This is the eighth instance of brushite occurrence on the territory of Czechoslovakia, and points to the relatively extensive occurrence of this mineral. The physical and optical properties of the indicated minerals were determined and qualitative analyses were run on them.

Card : 1/1

CZECHOSLOVAKIA / Physical Chemistry. Crystals. B

APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001547710011-4"

Abs Jour : Ref Zhur - Khimiya, No 12, 1959, No. 41391

Author : Sekanina, Josef

Inst : Not given

Title : Crystallographic Terminology

Orig Pub : Casop. mineral. a geol., 1958, 3, No 3,
345-349

Abstract : A list of 65 crystallographic terms in the Czech language is presented.

Card 1/1

SERAFIMOV, J.

"Crystallographic terminology."

CASOPIS PRO MINERALOGII A GEOLOGII., Praha, Czechoslovakia., Vol. 4, No. 1, 1959

Monthly List of EAST EUROPEAN ACCESSIONS (BEAI), LC, Vol. 6, No. 7, July 1959, Unclassified

SEKANINA, Josef

On the occasion of the 60th birthday of professor Jiri Novak. Vestnik
ust geolog 37 no.6:483 N '62.

CZECHOSLOVAKIA

SEKANINA, J.

Prague, Casopis pro mineralogii a geologii, No 2, 1963, pp
178-187

"Magnesia Skarn in Dolomite from Cichov in Western Moravia."

SEKANINA, Josef, prof., dr. (Sumavska 30, Brno)

Calcite from Stramberk; a morphologic and crystallogenic
study. Prace CSAV Brno 34 no.12:541-598 '62.

1. Clen korespondent Ceskoslovenske akademie ved.

SEKANINA, Iosif [Sekanina, J.]

Interference pleochroism. Min. sbor. no.16:387-391 '62.
(MIRA 16:10)

1. Universitet imeni Yana Ev. Purkine, Brno, Chekhoslovakiya.
(Crystal optics)

SEKANINA, Josef

Magnesia skarn in the dolomite near Cichov in western
Moravia. Cas min geol S no.2:178-188 Ap '63.

SEKANINA, Josef

On the ontogeny of calcite from Stramberk and on calcite
in general. Cas min geol 8 no.2:217-219 Ap.'63.

SEKANINA, Josef

Sixtieth birthday of Jaroslav Kokta. Vest ust geol 39 no.2:
153 Mr'64

SEKAKINA, Josef.

Nomenclature of geological sciences in Czechoslovakia and its
tasks. Vest. Mat. geol. 39 no.4: 303-306 '64.

CZECHOSLOVAKIA

SEKANINA, J.

J. E. Purkyne University (Universita J. E. Purkyne), Brno

Prague, Casopis pro mineralogii a geologii, No 3, 1964, pp 371
- 373

"Report on the Activity of the International Mineralogical
Association in 1962 and 1963."

SMEKANINA, Josef'

Report on the activity of the International Association of Mineralogy
in the years 1962 and 1963. Cas. Min. geol. 9 no. 3: 371-373 '64.

I. M. Příbram University, Brno.

SEKANINA, M.; SEKANINOVA, M.

Epidemic of diarrhea in a pediatric department caused by Pseudo-
monas aeruginosa. Cesk.pediat. 15 no.3:230-233 Mr '60.

l. Detske oddeleni OUNZ v Novem Meste na Mor., prim. MUDr.
J. Zemanek.

(PSEUDOMONAS INFECTIONS in infancy & childhood)
(DIARRHEA in infancy & childhood)

SEKANINA, M.

Complete systems of the neighborhood of sets in general topologic spaces.
c. 135. (SPISY, No. 374, 1956, Brno, Czechoslovakia)

SO: Monthly List of East European Accessions (EEL) LC, Vol. 6, No. 12, Dec 1957. Uncl.

SEKANINA, M.

Certain characteristics of compact related sets in Euclidean space.

p. 129 (CASOPIS PRO PESTOVANI MATEMATIKY) Vol. 82, no. 2, May 1957,
Praha, Czechoslovakia

SO: Monthly Index of East European Accessions (EEAI) LC, Vol. 7, No. 3,
March 1958

SEKANINA, Milan (Brno)

On a certain problem of a set of independent elements of an Abelian
group. Cas pro pest mat 85 no.3:338-341 Ag '60. (EEAI 10:1)
(Abelian groups) (Aggregates)

SEKANINA, Milan (Brno, Kotlarska 2)

A remark on factorization of non-commutative groups. Cas pro
pes mat 87 no.1:94-97 '62.

1. Prirodovedecka fakulta university J.E. Purkyne.

POLAK, Vatslav [Polak, Vaclav]; SEKANINA, Milan

On decomposition of the plane into system of subsets of topological circles. Cas pro pes mat 88 no.1:14-28 '63.

1. Prirodovedecka fakulta, Universita J.E.Purkyne, Brno,
Kotlarska 2.

SEKANINA, Milan

On an ordering of the vertices of a graph. Cas pro pest mat
88 no.3:265-282 Ag '63.

1. Prirodovedecka fakulta university J.E. Purkyne, Brno,
Kotlarska 2.

SEKANINA, Milan

On factorization of a set of nonnegative integers. Chekhosl mat
zhurnal 14 no. 2:161-170 '64.

1. Faculty of Natural Sciences, J.E.Purkyne University, Brno,
Kotlarska 2.

SEKANINA, M.

Four cases of arrhythmia in newborn infants. Cesk. pediat. 19
no. 7:627-630 Jl '64

1. Detske oddeleni nemocnice v Novem Meste na Morì vedouci:
MUDr. J. Zemanek.

SEKANINA, Milan

Systems of topology on a given set. Chekhosl mat zhurnal
15 no.1:9-29 '65.

l. J.E.Purkyne University, Brno, Janackovo nam.2A. Submitted
January 12, 1963.

SEKANINA, VLADIMÍR

V 2301* For Better Quality of Safety Razor Blades. Za lepší
jakost holíčich čepelek. (Czech.) Vladimír Sekanina. Hutičské
listy, v. 10, no. 11, Nov. 1955, p. 665-678.
MG Czechoslovakian methods for producing razor blades are com-
pared with procedures in other countries. Tight production
control measures are required to make a high quality product.
Tables, micrographs, diagrams, 5 ref.

WIEDERMANN, Milos, MUDr; SEKANINA, Zdenek, MUDr

Sheehan's syndrome. Ces.lek.cesk. 95 no.33-34:925-929 24 Aug 56.

l. I. interni klin. PU Olomouc--Predn. Prof. MUDr Pavel Lukl
M.W.I. int. klinika PU, Olomouc
(PITUITARY GLAND, ANTERIOR, dis.
Sheehan's synd. (Cz))

SEKANINA, Z.

SCIENCE

Periodicals: BIULETEN ASTRONOMICHESKIKH INSTITUTOV CHEKHO-SLOVAKII
BULLETIN OF THE ASTRONOMICAL INSTITUTES OF CZECHOSLOVAKIA.
Vol. 10, no. 2, Mar 1959

SEKANINA, Z. Variability of the photometrical exponent of the dust
part of the cometary coma. In English. p. 60.

Monthly List of East European Accessions (EEAI) LC, Vol. 8, No. 5,
May 1959, Unclass.

SEKANINA, Z.

Solar activity as related to the statistics of comets. In English.
p. 103.

BULLETIN OF THE ASTRONOMICAL INSTITUTES OF CZECHOSLOVAKIA. (Ceskoslovenska akademie
ved. Astronomicky ustav) Praha, Czechoslovakia, Vol. 10, no. 3, May 1959.

Monthly List of East European Accessions (EEAI), LC, Vol. 8, no. 11, Nov. 1959
Uncl.

SEKANINA, Z.; VAMYSEK, V.; GRYGAR, J.

Initial velocity in the tail of comet 1956h. In English. p. 115.

BULLETIN OF THE ASTRONOMICAL INSTITUTES OF CZECHOSLOVAKIA. (Ceskoslovenska akademie
ved. Astronomicky ustav) Praha, Czechoslovakia, Vol. 10, no. 4, July 1959.

Monthly List of East European Accessions (EEAI), LC, Vol. 8, no. 11, Nov. 1959
Uncl.

3.2440 (1060)

26844
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D244/D306

AUTHOR: Sekanina, Zdeněk (Prague)

TITLE: The study of physical properties of the cometary atmosphere

PERIODICAL: Pokroky matematiky, fyziky a astronomie, no. 1,
1961, 3 - 15

TEXT: The author summarizes achievements in the study of the atmosphere of comets to supplement a previous article on the physical structure of comets by V. Vanýsek (Pokroky matematiky, fyziky a astronomie 1 (1956), no.2, p. 156). The dependence of the absolute surface temperature of a comet's nucleus on the heliocentric distance, according to Markovič, can be expressed by the function $T(r) = T_0 \cdot r^{-\alpha}$, where r is the heliocentric distance; T_0 (the absolute temperature in a heliocentric distance of 1 AU) and α are parameters for which Markovič lists the following values given in Table 1. Some surface temperatures of cometary nuclei (ice + meteoric material conglomerate) and of

Card 1/9

26844
Z/028/61/000/001/001/002
D244/D306

The study of physical properties...

very fine dust particles under conditions of thermal equilibrium are listed in Table 2. Various studies indicated that the cometary atmosphere is not a homogeneous physical medium, and a so-called gas-dust model of a comet was proposed by Vanysek and collaborators. It is based on the assumption that the emission of a comet is caused by both gas molecules (fluorescence) and dust particles, in addition to the nucleus (which reflects the sunlight), so that the total photometric exponent (n), i. e. the instantaneous change of the total brightness of the coma in a given heliocentric distance, can be expressed by the two partial exponents (n_g) and (n_d). The correlation between the amount of molecules (N_o) liberated per time unit from a unitary surface area of the nucleus and its surface temperature is given by the Levin equation $n_o = N_o \cdot \left(\frac{\kappa T}{2\pi m} \right)^{1/2} \cdot e^{-\frac{L}{R_o T}}$, where N_o is the

molecule concentration in the surface layer of the nucleus, κ the Boltzmann constant, R_o the gas constant, T the absolute sur-

Card 2/9

26844
Z/028/61/000/001/001/002
D244/D306

The study of physical properties...

face temperature of the nucleus, and L the temperature required for the liberation of a certain amount of gas. There is still no definite answer as to the existence of dust in a comet's atmosphere. Sekanina found a combination of major parameters of dust contained in the atmosphere of a comet (so-called dust function) which allows the photometric exponent of the dust coma to be determined when all the parameters are known:

$$n_d = 2 - \frac{r}{\sqrt{ + \left(\frac{R}{\xi}\right)^2}} \cdot \frac{d\sqrt{\cdot}}{dr}$$

where R is the effective radius of the cometary nucleus, ξ the radius of the dust particles, and $\sqrt{\cdot}$ the total of the photometrically effective dust particles in the coma in a given heliocentric distance. Vanýsek calculated values for $\xi \sim 2 \cdot 10^{-5}$ cm, and assumes that the total amount of matter ranges from 10^{10} to

Card 3/9

26844
Z/028/61/000/001/001/002
D244/D306

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The study of physical properties...

10^{11} gram in the dust layer and from 10^{16} to 10^{21} gram in the nucleus. The Bessel model of the shape of a comet was replaced by another theory proposed by Orlov, who assumes that the movement of each particle (i. e. molecules and dust) is influenced by the following three forces: (a) gravitation of the Sun; (b) repulsion of the Sun (pressure of solar radiation); and (c) the pressure of the radiation reflected from the surface of the comet's nucleus. According to the Orlov model, the resulting distance from the peak of the comet's head to its nucleus is then

$\xi_0 = (2r)^{\frac{1}{2}} \cdot \left(\frac{\mu_1}{1 + \mu} \right)^{\frac{1}{2}}$ where $1 + \mu$ is the repulsion of the Sun, and μ_1 the effective acceleration per molecule in direction from the nucleus. Orlov also assumes that the acceleration effected by the Sun and the comet's nucleus are proportional to the illumination produced by the two bodies and formulates:

Card 4/9

26844
Z/028/61/000/001/001/002
D244/D306

The study of physical properties...

$$\frac{10^{-0.4} h_0}{10^{-0.4} H_\odot} = \frac{\mu_1}{1 + \mu}$$

an expression which allows the stellar size of the nucleus (h_0),
to be determined since the stellar size of the Sun (H_\odot) is known
and the ratio of the two accelerations can be derived by observ-
ing the motion of peaks (so-called envelopes) in the coma. Orlov
also found that one comet can have up to four such envelopes which
always have definite distances from the nucleus in the ratio $\xi_1:$
 $\xi_2 : \xi_3 : \xi_4 = 1 : 2.4 : 3.9 : 5.6$. From the deformation
and irregularities of envelopes observed in the coma of the More-
house 1908 III comet, Orlov concluded that the gas emission from
the nucleus is not continuous, but occurs in form of explosions.
This is contradicted by Dobrovolskiy, who stated that the enve-
lopes in the coma of the Morehouse comet contained ionized CO^+
(so that the electromagnetic forces must be considered in the

X

Card 5/9

26844

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D244/D306

The study of physical properties...

calculation) and that envelopes are not exclusively gaseous formations, but contain also dust particles. According to Dobrovolskiy, so-called parent molecules (M) are liberated from the dust particles. These molecules do not radiate themselves, but they dissociate rapidly under the influence of solar corpuscular radiation: $H^+ + M \rightarrow H + M^+$; neutral CO is the parent molecule and electrons liberated by its ionization are dragged into the stream of solar corpuscular radiation, thus creating an electrostatic field which acts on the luminous CO^+ ions. Halos, sometimes observed in the heads of comets are caused by larger amounts of particles, ejected from the nucleus into all directions. Precise studies on the geometry of this phenomenon were made by Mochnač. The author describes now the physical development of the atmosphere of a comet: At large distances from the Sun, the surface temperature of the cometary nucleus is very low (-100°C and less) and the comet remains practically inactive. As the comet approaches the Sun, the surface of the nucleus is heated and the first gases are liberated at Sun-comet distances of 3 -

Card 6/9

26844

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D244/D306

The study of physical properties...

2 AU. The extension of the comet increases rapidly and the entire area which can be filled with gases is "saturated" at Sun-comet distances of 1-2 AU. Three individual stages are distinguished in the development of a cometary atmosphere: (1) Until to the moment when intensive evaporation of frozen gases sets in; (2) Until to the moment when the atmosphere is saturated; and (3) The stage of the "saturated" coma. The most important parameters of the development are heliocentric "limit" distances which, in turn, are influenced by the dimensions and composition of the cometary nucleus. Comets can also be classified as "old" and "new" according to the number of orbits around the Sun, since the lifetime of comets depends on the length of orbiting periods. The best evidence for determining the age of a comet is the curve of its atmospheric development (i. e. its "limit" heliocentric distances). However, such curves are as yet established only for few comets. In conclusion, the author states that theoretical studies of processes occurring in the coma and improved observation instruments will greatly contrib-

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Card 7/ 9

26844

Z/028/61/000/001/001/002
D244/D306

The study of physical properties...

ute to the further development of cometary physics. *[* Abstrac-
ter's note: Papers of cited authors are not listed *]*. There
are 5 tables and 5 figures.

Předpokládaná struktura jádra	Před periholem 4)		Po perihelu 5)	
	T_0	α	T_0	α
led H ₂ O 2)	157°K	0,23	162°K	0,18
konglomerát ledu H ₂ O + těžko tavitelného meteorického materiálu 3)	161°K	0,18	160°K	0,17

Table 1 Legend: 1) Assumed structure of the comet's nucleus
2) Conglomerate of H₂O ice and difficultly fusible
meteoric material 3) H₂O ice 4) Before perihelion
5) After perihelion

Card 8/9

S/035/62/000/004/021/056
A001/A101

AUTHOR: Sekanina, Z.

TITLE: Visual effects following the landing of Lunik II on the Moon

PERIODICAL: Referativnyy zhurnal, Astronomiya i Geodeziya, no. 4, 1962, 64,
abstract 4A536 ("Ríše hvězd", 1961, v. 42, no. 2, 27-30, Czech)

TEXT: This is a summary and survey of visual effect observations connected
with the landing on the Moon's surface of the second Soviet space rocket on
September 14, 1959. It is compiled mainly on the basis of data in the article
by O. B. Dluzhnevskaya (RZhAstr, 1961, #12A591). ✓

V. B.

[Abstracter's note: Complete translation]

Card 1/1

*S/035/61/000/012/027/043

SEKANINA, Z.

Secular variations in the absolute brightness of short-period comets. Biul astr Cz 15 no.1:1-7 '64.

Perihelion asymmetry of photometric curves of comets.
Ibid.:8-20.

l. Public Observatory, Prague.

L 45085-66

ACC NR: AP6026464 SOURCE CODE: CZ/0092/66/017/002/0067/0084

AUTHOR: Sekanina, Z.

ORG: Public Observatory, Prague

TITLE: New original and future comet orbits

SOURCE: CSAV. Byul astron inst Chekhoslov, v. 17, no. 2, 1966, 67-84

TOPIC TAGS: comet, comet orbit/Zuse Z23 digital computer

ABSTRACT: The author discusses current methods used to determine original and future comet orbits, and computes the original and future orbits for four comets of the 1940's and 1950's according to a method of his choice. They are namely: 1949 I (Wirtanen), 1949 IV (Bappu-Bop-Newkirk), 1953 III (Mrkos-Honda) and 1958 III (Burnham). The author also establishes the future orbits for comets 1905 IV (Kopff), 1905 VI (Brooks) and 1946 VI (Jones). A brief description of each comet follows and the results of the computations are given in the form of tables in the original article. In addition, the value of the auxiliary-anomaly coefficient of Makover's method [Makover, S. G., 1955, BIA 6, 244] is discussed

14
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12

Card 1/2

L 45085-66

ACC NR: AP6026464

as a function of the heliocentric distance of the perihelion. The author thanks
Miss I. Strakova for her careful tape-punching for the Zuse Z23 digital computer.
Orig. art. has: 14 formulas and 11 tables. [GC]

SUB CODE: 03, 09 / SUBM DATE: 13Sep66 / SOV REF: 008 / OTH REF: 011 /

Card 2/2 b1g

JANICEK, M.; SEKANINOVA, H.; VACLAVSKY, J.

Relation of smoking to the incidence of hypertension. Cas. lek.
cesk. 103 no. 24:659-662 12 Je'64

1. I. vnitri klinika lekarske fakulty PU [Palackeho university]
v Olomouci; prednostas prof. dr. P. Lukl.

JANICEK,M.; KUBIS,M.; JANOUSEK, M.; SEKANINOVA, H.

Effect of obesity and some dietary components in the pathogenesis
of hypertension. Vnitrní lek. 11 no.2:130-138 F '65

1. I. vnitrní klinika Palackého University, Olomouc (prednosta:
prof. MUDr. P. Luk.).

ROZELKA, J.; ROMY, J.; CHALOUKOV, M.; PAVLICK, J.

The significance of epizootic disease in cattle carriers
to infectious mastitis. Veterin. Let. 11 n. 2:37-38.
April.

J. Rozelka (Veterinary Faculty of Masaryk University (presi-
dence prof. Dr. Pavel, DVM); Veterin. Faculty of Masaryk
University v Olomouci (president: prof. Dr. Bohus Salcher, DVM)

WONDRAK, Ed.; SEKANINOVA, J.

On treatment of fractures of the tibial condyle. Acta chir. orthop.
trauma. Cech; 28 no.6:491-498 D '61.

1. Chirurgicka klinika lekarske fakulty Palackeho university v
Olomouci prednosta prof. MUDr. Vlad. Rapant, Dr. Sc. Rehabilitacni
oddeleni fakultni nemocnice v Olomouci, prednosta MUDr. Karel
Tabarka, C.Sc.

(TIBIA fract & disloc)

SEKARDI, Ljerka; EHRILCH, Ivo

Acetylcholinesterase of *Fasciola hepatica* L.
Biol glas 15 no. 4: 229-233 '62.

1. Institut za biologiju Sveucilista u Zagrebu.

SEKARDI, M.

Cooling water supplies in chemical industry. M. Sekardi (Neft, Zagreb, 1954, 6, 63-19).—Needs, sources, and uses of cooling water in different branches of chemical industry are discussed. Methods of re-cycling with cooling in spray towers are described and some design problems are outlined.

S. K. LACHOWICZ

11(4)

AUTHOR: Sekardi, Marijan, Engineer YUG/4-59-1-1/16

TITLE: Columns With Bubble-Cap Plates. I. (Kolone sa zvona-
stim pliticama. I)

PERIODICAL: Nafta, 1959, Nr 1, pp 1-10 (YUG)

ABSTRACT: The article gives detailed description of the me-
chanical composition and function of fractionating
columns with bubble-cap plates. Calculations of
various factors which influence the efficiency of
bubble-cap plates are based on English and American
methods and formulas. There are 8 graphs, 3 dia-
grams and 9 references of which 7 are English, 1
German and 1 Yugoslav.

ASSOCIATION: Institut za naftu - Zagreb. (Institute for Petroleum
- Zagreb)

~~SECRET~~, M.

Towers with sieve trays. p. 231.

ANTA. (Institut za naftu)
Zagreb, Yugoslavia
Vol. 10, no. 7, July 1959.

Monthly List of Eastern European Accession Index (EELI) ED vol. 1, No. 11
December 1959
Encl.

SEKATSKIY, A.

How we learned to use the "Donbass" cutter-loader. Mast.ugl. 5
no.4:7-8 Ap '56. (MIRA 9:7)

1.Nachal'nik uchastka shakhty No.9 "Lipkovskaya" kombinata
Tulaugol'.
(Moscow Basin--Coal mining machinery)

KOLAR, Z.; DEVELIC, Gj.; RANDIC, M.; TRINAJSTIC, N.; SEKE, V.

Book reviews. *Croat chem acta* 35 no.4:315-319 '63.

l. Clan Redakcionog odtora, "Croatica Chemica Acta" (for
Randic).

SEKE, Vesna

Means for the increase of viscosity of photographic emulsions. Kem
ind 10 no.1:F-18--F-21 Ja '61.

1. "Fotokemika", Zagreb.

ACC NR: AP6032806

SOURCE CODE: YU/0001/66/000/009/1594/1600

AUTHOR: Seke, Vesna (Graduate engineer; Consultant); Fles, Dragutin (Doctor; Director)

ORG: Organic Chemistry Industry, Zagreb (Organsko kemijska industrija)

TITLE: Study of optically active polyamides. II. Synthesis of optically active poly-(γ -methyl- ϵ -caprolactam [First part appeared in Croatica Chimica Acta, 1966, No. 1, entitled: "The Absolute configuration of γ -methyl- ϵ -caprolactam"]]

SOURCE: Tehnika, no. 9, 1966, 1594-1600

TOPIC TAGS: polymer, polymerization, monomer, optically active polymer, polyamide

ABSTRACT: A survey of methods for the preparation of optically active polymers is given. In the experimental part the polymerization of optically active and racemic γ -methyl- ϵ -caprolactam from (S) (+) γ -methyl- ϵ -caprolactam and (\pm) γ -methyl- ϵ -caprolactam is described. (S) (+) γ -methyl- ϵ -caprolactam is obtained by two different methods: the first by conversion of optically active

Card 1/2

UDC: 678.675=861

ACC NR: AP6032806

α -methyl- γ -phthalimidobutyric acid to γ -methyl- γ -phthalimidocaproic acid through two Arndt-Eistert syntheses; the second, by the resolution of γ -methyl- γ -phthalimidocaproic acid via diastereomeric quinine salt. The first method was used for determination of the absolute configuration of γ -methyl- γ -caprolactam, while the second was applied as a convenient preparative method for the preparation of optically active caprolactam. Polymerization of racemic and optically active monomers of high purity was carried out at 210°C for 90 hr using 1% of water as a catalyst. Polymer yields were 59% for the optically active monomer, and 41.5% for the racemic monomer. Infrared spectra, intrinsic viscosity, and the influence of the solvent's composition on the specific rotation is given. Specific rotation of (S) (+) poly-(γ -methyl- γ -caprolactam) showed a sharp maximum at 30 vol % of chloroform in m-cresol. Orig. art. has: 4 figures. [Based on authors' abstract]

[KS]

SUB CODE: 07/SUBM DATE: none/ORIG REF: 001/SOV REF: 002/OTH REF: 031

Card 2/2

SEKE, Vesna (Zagreb)

Fixing baths. Kem ind 10 no.7; Suppl. F-71--F-80 J1 '61.

VAN DORMAEL, A.; NYS, J.; DEPOORTER, H.; SEKE, V. [translator]

New group of sensibilizers for photographic emulsions. Kem
ind 10 no. 11; 443-444 N '61.

SEKEY

SABO, I.; MODI, Y. [Modi, I.]; SEKEY, Ya. [Sekey, I.] (Tyrgu-Muresh,
Rumyniya)

Effect of silicic acid and silica on blood lipoids. Pat.fiziol.
i eksp. terap. 5 no.3:74-75 My-Je '61. (MIRA 14:6)

1. Iz nauchno-issledovatel'skoy bazy (rukovoditel' - akademik
D. Mishkol'tsi) Akademii nauk Rumynskoy Narodnoy Respubliki
i kafedry fiziologii (rukovoditel' - dotsent I.Sabo) Mediko-
farmatsevticheskogo instituta.
(LIPIDS) (SILICA)

SANTO D'YERD' [Santo György]; SMKEL', Otto

Severe injuries of the chest with disturbance of its function and treatment by prolonged artificial respiration with the aid of respiratory apparatuses. Khirurgiia 35 no.10:79-84 O '59.

(MIRA 12:12)

1. Iz Gosudarstvennogo instituta travmatologii (dir. - doktor Santo D'yerd'), Budapest.

(THORAY wds & inj.)

(RESPIRATION, ARTIFICIAL)

SEKELJ, Antonije, pukovnik

Protection of medical installations from ground and air attack.
Voj. san. pregl., Beogr. 11 no.11-12:697-700 Nov-Dec 54.

1. Katedra vojnih i vojnosanitetskih predmeta VMA.

(CIVIL DEFENSE

med. installations, protection from ground & air attack)

(MEDICINE, MILITARY AND NAVAL

med. installations, protection from ground & air attack)

SEKER D

✓ Investigation of current efficiency on electrolysis of cryolite-alumina melts. D. Seker and A. I. Il'yayev, *Sbornik Nauch. Trudov Naukov. Inst. Tsvetnykh Metal.*, Zoloto 1954, No. 24, 117-30; *Referat. Zhur., Met.* 1956, No. 1112. — The influence of various factors on the loss of Al and current efficiency (c.e.) on electrolysis of cryolite-alumina melts (1) was investigated under lab. conditions. Min. loss of Al and max. c.e. corresponds to cryolite ratio 2.7. Introduction into the electrolyte of 5 wt. % CaF₂ or MgF₂ raises the c.e. by 1-2% or 0-7%, resp., on account of reduction of loss of Al because of increase in surface tension of electrolyte at the interface with molten Al under the influence of Ca⁺⁺ and Mg⁺⁺. In the presence of 5% CaF₂ or MgF₂ max. c.e. corresponds to a cryolite ratio of 2.5. Up to 10 wt. % CaF₂ or MgF₂ c.e. at still higher concn. c.e. rapidly decreases as a result of increasing the sp. gr. and cryolite ratio with increasing the max. loss of Al goes through a maximum. With 10 wt. % CaF₂ or MgF₂ Al loss reaches a minimum, c.e. reaches a maximum. At min. c.e. there is a sharp increase in the loss of Al. Out of the factors influencing the loss of Al, the temp. of electrolyte, increase of current density, and distance, c.e. increases. Al losses decrease, while the concn. in anode gases increases. At const. temp., inter-electrode distance, and c.e. concn. of CO₂ in anode gases varies directly as the concn. of Al₂O₃ in the electrolyte and can serve as a c.e. indicator. To insure max. c.e. it is expedient to work at max. possible concn. of Al₂O₃ in the electrolyte of a given compn.; this requires continuous feeding of baths with Al₂O₃.

V. N. Bednarek

10

The development of the nomenclature of organic compounds. Aleš Sekera, *Chemic* (Prague) 4, 2-3 (1948).
S extends the application of the simplified Dyson system
of naming org. compds. Frank Maresch

CA

The Raney catalysts. Aleš Sekera. *Chemie* (Prague) 5, 5-7(1949).—In hydrogenation the use of the catalysts Co, Fe, Cu instead of Ni does not permit uniform conclusions. These catalysts require a longer reaction time, higher temp., and higher pressures, but they produce fewer side reactions than does Ni and consequently yield purer products.

Frank Maresh

CA

17

Anesthetics of carbamic acid series... A. Sekera, J. Hrubý, Č. Vrba, and J. Lebdulka (Masaryk Univ., Brno, Czech.). *Chem. Listy* 44, 275-6 (1950).—From excess RNCO and Et₂NCH₂CH₂OH, the following RNHCOOCH₂CH₂NEt₂ compds. were obtained by heating in toluene, filtering off the disubstituted urea after a few days standing, and pptg. the product with dry HCl in the form of its HCl salt in 80-90% yield. B.ps. of RNCO and m.ps. of RNHCOOCH₂CH₂NEt₂.HCl are given for the different R values Ph (b, 48-9°, m. 146.5°), 1-naphthyl (b, 130-1°, base m. 80.5°, picrate 134°), 2-naphthyl (m. 56.5°, m. 160°, picrate 191.5°), o-tolyl (b₁, 63-4°, m. 168.5°), m-tolyl (b₁ 55-7°, m. 134.5°), p-tolyl (b₁ 49-51°, m. 186°), 2,4-dimethylphenyl (b₁ 76-8°, m. 161°), p-MeOC₆H₄ (b₁ 98-100°, m. 171.5°), p-EtOC₆H₄ (b₁ 100-1°, m. 173°). In addn., Ph₂NCOOCH₂CH₂NEt₂ (m. 180°) was prep'd. from Ph₂NCOCl (m. 84°). Anesthetic action was 0-0.3 as compared to 1 of cocaine. M. Hudlický

SEKERA, A.

Certain new problems in the investigation on local anesthetics. Lek.
listy, Brno 6 no.19:565-570 1 Oct 51. (CLML 21:4)

1. Of the Institute of Pharmacological Chemistry of Masaryk University
Brno.

c/t

17

Anesthetics from carbamic acid series. II. A. Sekera,
A. Borovanský, and Č. Vrba (Masaryk Univ., Brno,
Czech.). *Chem. Listy* 45, 90 (1951); cf. *C.A.* 43, 3122f.—
By the reaction of $\text{Me}_2\text{N}(\text{CH}_2)\text{OH}$ with alkyl isocyanates in
ether, the following derivs. of $\text{RNHCOOCH}_2\text{CH}_2\text{NMe}_2$ were
prepd. and characterized by the m.p. of the HCl salts:
 Me , m. 117-19°; Et , m. 110°; Pr , m. 89-90°; Bu , m.
97-98°; iso-Pr , m. 126°; iso-Am , m. 112-14°; tert-Bu , m.
137°. No anesthetic activity was noted. M. Hudlický

1951

SEKERA, A.; RAHM, J.

Antural altihelminthic drugs; askaridol. Cesk. farm. 1 no.8:471-477
Sept 1952. (CIML 23:2)

VRBA, C.; LERDUSKA, J.; SEKERA, A.

Studies on local anesthetics; pharmacological evaluation of active
basic esters of substituted carbamic acids. Cesk. farm. 1 no.10:554-
563 1952.
(CIML 23:4)

1. Of the Institute of Pharmacology of the Veterinary School and of
the Institute of Pharmaceutical Chemistry of Masaryk University, Brno.

SEKERA, A.; RAHM, J.

Natural anthelmintics; santonin, Cesk. farm. 1 no. 10:594-598 1952.
(CML 23:2)

Author: A.

Studies in local anaesthesia. II. Synthesis of further basic esters of substituted carboxylic acids. 1952. (Chemicke Listy. Praha. Vol. 46, No. 12, Dec. 1952)

CC: Kralik, Listy of Czechoslovak Academies, (ASCR), LC, Vol. 4, No. 4, June 1953, p. 10.

SEKERA, A

SEKERA, A.; RAHM, J.

Natural antihelminthics, III. Helenin. Cesk. farm. 2 no. 1:22-24
Jan 1953. (CIML 25:1)

SEKERA, A.

SEKERA, A.; HRUBY, J.; JARUBEC, I.; KRAL, J.; VRBA, C.; LEBDUSKA, J.

Local anesthetics. Basic esters of substituted carbamic acids [with summary in English]. Sbor.Chekh.khim.rab. 18 no.6:870-879 D '53.
(MIRA 7:6)

1. Department of Pharmaceutical Chemistry of the University, and Pharmacological Department of the Veterinary School, Brno.
(Anesthetics) (Carbamic acid) (Esters)

SEKERA, ALES

Local anesthetics. III. Basic esters of monoalkylcarbamic acids. Aleš Sekera, Alois Borovanský, and Čeněk Vrba (Masarykova Univ., Brno, Czech.). *Chem. Listy* 47, 691-7 (1953); cf. *C.A.* 47, 12302e. — 2-Diethylaminomethyl N-alkylcarbamates (I) prep'd. from alkyl isocyanates or by the Curtius degradation were inactive as local anesthetics. Some of them showed a slight sedative and hypnotic action upon white mice. Dialkyl sulfates and KOCN gave MeNCO, b. 42-5° (42%), and EtNCO, b. 59-61° (36%). BuI (0.25 mole) refluxed 20-50 hrs. with 45 g. AgOCN in 200 ml. Et₂O gave 30% BuNCO, b. 112-13°; Me(CH₃)₂CH:CH-(CH₃)NCO (II), b. 102-3°, was prep'd. in 85% yield by boiling 2 hrs. 30 g. Me(CH₃)₂CH:CH-(CH₃)COCl in 100 ml. C₆H₆ with 13 g. NaN₃. The I were prep'd. by refluxing 0.25 mole RNCO 8-10 hrs. with 35.2 g. Et₂NCH₂CH₂OH (III) in 100 ml. Et₂O and distig. the mixt. *in vacuo*. The bases with HCl gas in Et₂O gave the HCl salt, crystd. from Et₂O-Me₂CO. Alkyls, % yields, and b.p.s. of the I, and m.p.s. of the HCl salts: Me, 62, b. 120°, 117-19°; Et, 77, b. 103°, 110°; Pr (IV), 20, b. 116° (n_{D}^{20} = 1.4000), 91-2°; Bu, 90, b. 128°, 97-9°; iso-Pr, 41, b. 110°, 128°; sec-Bu, 20, b. 122°, 112-14°; Me₂C, 25, b. 109°, 157°. CH₃:ClCH₂-Br and AgOCN gave CH₃:CHCH₂NCO which yielded 26% CH₃:CHCH₂NHCO₂CH₃NET₃, b. 121-3° (HCl salt, m. 126-7°), hydrogenated in EtOH over 20% Pd-C, to a compd. b. 135-40°, the consts., with the exception of the b.p., and the infrared spectrum of which agree with those of IV. Heating 5 g. III with 5 g. MeCH₂CON₃ in Et₂O gave 19% iso-BuNHCO₂CH₂CH₂NET₃, b. 148° (HCl salt, m. 120-8°). Refluxing 8 g. II and 5 g. III 5 hrs. in 30 ml. C₆H₆ gave Me(CH₃)₂CH:CH(CH₃)₂NHCO₂CH₂CH₂NET₃ (V); HCl salt, m. 103-10° (28% yield). Hydrogenation of V gave C₆H₅NHCO₂CH₂CH₂NET₃; HCl salt, m. 116-19° (from petr. ether-CHCl₃). M. Hudlický

Sekera, A

2597. New method of determining the melting-point of hygroscopic substances. A. Sekera and J. Pokorný (*Milrochim. Acta*, 1954, 3-4), 266-270.—The capillary m.p. tube has a small bulb containing P_2O_5 . The substance of which the m.p. is to be found is introduced into the tube past the P_2O_5 , and the tube is evacuated. After sufficient time to allow complete drying, the tube is sealed at a place beyond the P_2O_5 bulb, which is drawn off. The m.p. is then found in the usual way. The m.p. of nine substances determined by this method were higher than those obtained by the usual procedure. A. J. Max

SEKERA, Ales; POKORNY, Jaroslav

Kofler's block and its use in pharmacy; I, description and use of
the apparatus. Cesk.farm. 4 no.3:146-151 Apr 55.

(MICROSCOPY, apparatus and instruments,

Kofler's block for micromeasurement of melting point,
use in pharmacy)

SEKERA, Ales; KRACMAR, Josef

Use of Kofler's micromethod in drug control, I; identification
of chemical official drugs in Czechoslovakian pharmacopeia I.
Cesk. farm. 4 no.4:173-180 May 55.

1. Z ustanu pro farmaceutickou chemii Masarykova univerzity v
Brne.

(PHARMACOPEIA

Czech. I, drug identification by Kofler's
micromethod)

(DRUG INDUSTRY

identification by Kofler's micromethod)

SEKERA, Ales; POKORNY, Jaroslav

Kofler's block and its use in pharmacy; 1, determination of melting point. Cesk. farm. 4 no.4:194-198 May 55.

(MICROSCOPY

Kofler's block, micromeasurement of melting point, use in pharmacy)

SEKERA, Ales; KRACMAR, Josef

Use of Kofler's micromethod in drug control. II. Identification
of chemical official drugs in Czechoslovakian pharmacopeia 1,
suppl. 1. Cesk. farm. 4 no.5:232-233 June 55.

1. Z ustanu pro farmaceutickou chemii Masarykovy university v
Brne.

(PHARMACOPHIA

Czech. 1, supplement 1, drug identification by
Kofler's method)

(DRUG INDUSTRY

control of drugs by Kofler's micromethod)

BOROVANSKY, Alois, SEKERA, Ales

Glucochloroloses. I. Preparation of α - and β -glucochloralose.
Cesk.farm. 4 no.6:292-293 J1 '55.

1. Z Ustavu pro chemii farmaceutickou Masarykovy university v
Brne.

(HYPNOTICS AND SEDATIVES, preparation of,
 α - & β -glucochloralose)

SEKERA, Ales; STRANIK, Jan

Glucochloraloses. II. Electrophotometric determination of alpha-
and beta glucochloralose. Cesk. farm. 4 no.7:330-333 Sept 55.

1. Z Ustavu pro chemii farmaceutickou Masarykovy university v
Brne.

(HYPNOTICS AND SEDATIVES, determination,
 α - & β -glucochloralose, electrophoresis)
(ELECTROPHORESIS,
of α - & β -glucochlralose)

KUCA, Libor; BOROVANSKY, Alois; SEKERA, Ales

Glucochloralose. 3. Determination of β -glucochloralose in the presence of α -glucochloralose by spectrophotometry in the infra-red region. Cesk. farm. 4 no.8:412-414 Oct 55.

1. Z Ustavu pro chemii farmaceutickou Masarykovy university v Brne.

(HYPNOTICS AND SEDATIVES

α - & β -glucochloralose determ. by spectrophotometry in infra-red region)

(SPECTROPHOTOMETRY

determ. of β -glucochloralose in presence of α -glucochloralose in infra-red region)

(INFRA-RED RAYS

spectrophotometric determ. of β -glucochlralose in presence of α -glucochlralose)

SeKera, A.

Stability and local anesthetic effect in some derivatives of xylocaine. A. Sekera, J. Sová, and C. Vrba (Masaryk Univ., Brno, Czech.). *Experientia* 11, 276-0 (1965) [in English].—Several analogs of ω -diethylamino-2,6-dimethylacetanilide (I) (xylocaine) were studied with respect to the strength of the amide bond to test the hypothesis that the anesthetic effect of the I-type compds. is a function of the strength of these bonds. The following rate consts. ($\text{sec}^{-1} \times 10^{-4}$) were found for hydrolysis in 5*N* HCl at 89.5° with 0.01*M* initial substrate concn.: carbanilic acid, 2-diethylaminocethanol ester (II) 8.22; 2-methylcarbanilic acid, 2-diethylaminocethanol ester, 3.18 (III); 2,6-dimethylcarbanilic acid, 2-diethylaminocethanol ester, 0.33 (IV); ω -diethylaminoacetanilide, 81.5 (V); ω -diethylamino-2-methylacetanilide, 26.4 (VI) and I, 0.727. The order of effectiveness in surface anesthesia (rabbit cornea) in comparison to 0.01*M* cocaine was II, I, III, IV, V-VI; for infiltration anesthesia (guinea pig) V, VI, III, II, I, IV compared with 0.02*M* procaine. D.-S.E.